

Previous Year Paper

Chemistry - 2002



Exam Year 2002

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- 1. The first artificial disintegration of an atomic nucleus was achieved by
 - A. Geiger
 - B. Wilson
 - C. Madam Curie
 - D. Rutherford

Answer

- Irrespective of the source, pure sample of water always yields 88.89% mass of oxygen and 11.11mass of hydrogen. This is explained by the law of;
 - A. conservation of mass
 - B. constant composition
 - C. multiple proportion
 - D. Gay-Lussa

Answer

- 3. When the product of pressure and volume is plotted against pressure for a given amount of gas, the line obtained is :
 - A. parallel to X-axis
 - B. parallel to Y-axis
 - C. linear with positive slope
 - D. linear with negative slope

Answer

- 4. 32g of O₂, 2g of H₂ and 28g of N₂ at STP, occupy separately a volume of:
 - A. 1L
 - B. 2L
 - C. 22.4L
 - D. 2.24L

Answer

- 5. Air at sea level is dense. This is a practical implimentation of :
 - A. Boyle's law
 - B. Charles' law
 - C. Avogadro's law
 - D. Dalton's law

Answer

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B. sp² -hybridization

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- C. sp³ hybridization
- D. sp³d -hybridization

Answer

- 7. Oxygen molecule contains:
 - A. no unpaired electron
 - B. one unpaired electron
 - C. two unpaired electrons
 - D. three unpaired electrons

Answer

- 8. The law of thermodynamics formulated by Dr.N. Nernst is:
 - A. first law of thermodynamics
 - B. second law of thermodynamics
 - C. third law of thermodynamics
 - D. both (a) and (b)

Answer

- 9. Which of the following will favour the reverse reaction in a chemical equilibrium ?
 - A. Increasing the concentration of the reactants
 - B. Removal of at least one of the products at regular intervals
 - C. Increasing the concentration of one or more of the products
 - D. Increasing the pressure

Answer

10. K_0 and K_c are related as :

A.
$$K_p = K_c(RT)^{\Delta n}$$

B.
$$K_p + K_c = (RT)^{\Delta n}$$

C.
$$K_c = K_p$$

D.
$$K_n . K_c = (RT)^{\Delta n}$$

Answer

- 11. The pH of $0.005 \text{ M} \text{ H}_2\text{SO}_4$ is :
 - A. 2.5
 - B. 1.5
 - C. 1.0
 - D. 2.0

Answer

- 12. Lassaigne's test is used to detect :
 - A. nitrogen and halogens

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C. halogens and sulphur

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Answer

- 13. The number of possible structural isomers of butene are:

 - B. 2
 - C. 4
 - D. 1

Answer

- 14. Disymmetric object is one which is:
 - A. superimposable on its mirror image
 - B. non-superimposable on its mirror image
 - C. optically inactive
 - D. achiral

Answer

- 15. Cycloalkane has the formula:
 - A. C_nH_{2n+2}
 - B. C_nH_{2n-2}
 - C. C_nH_{2n}
 - D. $C_{2n}H_2$

Answer

- 16. Three fused benzene rings are found in:
 - A. naphthalene
 - B. anthracene
 - C. phenanthroline
 - D. triphenyl methane

Answer

- 17. To differentiate between carbon-12, carbon-13 and carbon-14, the instrument that you would use is:
 - A. infra-red spectrometer
 - B. atomic absorption spectrometer
 - C. mass spectrometer
 - D. ultraviolet spectrometer

Answer

- 18. A mixture of sand and iodine can be separated by:
 - A. crystallisation
 - B. sublimation

C. distillation Like. Share. Bookmark. Download. Make Notes. Print - Your Favourite Questions. Join www.zigya.com



Answer

- 19. Pyrostilicate forms, Solved Previous Year Papers . Questions and Answers. Free Forever.
 - A. SiO22-
 - B. Si2O76-
 - C. SiO42-
 - D. SiO3-

Answer

- 20. The effective component, of bleaching powder.....is the of calcium :
 - A. Cl
 - B. OCI
 - C. O²⁻
 - D. O₂Cl⁻

Answer

- 21. In the lime kiln, the reaction $CaCO_3(s) \rightarrow CaO(s) + CO_2(g)$ goes to completion because:
 - A. of high temperature
 - B. CaO is more stable than CaCO₃
 - C. CO₂ escapes simultaneously
 - D. CO₂ is a gaseous product

Answer

- 22. The rate of a reaction is doubled for every 10° rise in temperature. The increase in reaction rate as a result of temperature rise from 10° to 100° is :
 - A. 112
 - B. 512
 - C. 400
 - D. 614

Answer

- 23. When a beam of light is passed through a colloidal solution it :
 - A. is reflected
 - B. is scattered
 - C. transmitted
 - D. absorbed

Answer

- 24. The electron affinity values for the halogens show the following trend :
 - A. F < Cl > Br > I
 - B. F < Cl< Br< I
 - C. F > Cl > Br > l

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25. When H_2S gas is passed through nitric acid, the product is:

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- B. amorphous S
- C. monoclinic S
- D. plastic S

Answer

- 26. Atoms in a P_4 molecule of white phosphorus are arranged regularly in space in which of the following way ?
 - A. At the comers of tetrahedron
 - B. At the comers of a cube
 - C. At the comers of a four membered ring
 - D. At the centre and comers of a equivalent triangle

Answer

- 27. The catalytic activity of the transition metals and their compounds is described to :
 - A. their chemical reactivity
 - B. their magnetic behaviour
 - C. their unfilled d-orbitals
 - D. their ability to adopt multiple oxidation states and their complexing ability

Answer

- 28. The compound ZnFe₂O₄ is:
 - A. a normal spinel compound
 - B. a inverse spinel compound
 - C. interstitial compound
 - D. covalent compound

Answer

- 29. High purity Si and Ge for semiconductor properties can be obtained by :
 - A. calcination
 - B. roasting
 - C. zone refining
 - D. thermic process

Answer

- 30. Name of the alloy of aluminium which is used in aeroplane is :
 - A. duralumin
 - B. bell metal
 - C. aluminium bronze
 - D. alumina

Answer

31. Invar, an alloy of Fe and Ni is used in watches and meter scale. Its characteristic property is Like. Share. Bookmark. Download. Make Notes. Print - Your Favourite Questions. Join www.zigya.com

B. resistance of corrosion

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D. magnetic nature

Answer

- 32. An alloy of Pb and Sn in equal proportion is called:
 - A. solder
 - B. constantan
 - C. gun metal
 - D. type metal

Answer

- 33. Aqua-regia is obtained by mixing two different acids in the ratio :
 - A. 1HNO₃: 3HCl
 - B. 2HNO₃: 2HCl
 - C. 1HNO₃: 2HCl
 - D. 3HNO₃: 1HCl

Answer

- 34. A compound with cubic structure is made of elements A and B. A atoms are at the corners of the cube and B atoms are at the face centres. The simplest formula of the compound is :
 - A. A₅B
 - B. AB₃
 - C. AB
 - D. AB₆

Answer

- 35. Colligative properties are used for the determination of :
 - A. molar mass
 - B. equivalent weight
 - C. arrangement of molecules
 - D. melting point and boiling point

Answer

- 36. Identify the mixture that shows positive deviation from Raoult's law:
 - A. $CHCl_3 + (CH_3)_2CO$
 - B. $(CH_3)_2CO + C_6H_5NH$
 - C. $CHCI_3 + C_6H_6$
 - D. $(CH_3)_2CO + CS_2$

- 37. If ΔG for a reaction is negative, you infer that the change is :
 - A. spontaneous

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C. reversible

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Answer

- 38. $Zn(s) | Zn^{2+}(aq)| | Cu^{2+}(aq) | Cu(s) is :$
 - (anode)

(cathode)

- A. Weston cell
- B. Daniel cell
- C. Faraday cell
- D. Standard cell

Answer

- 39. An important industrial solvent, also a laboratory reagent called 2-butanone is nothing but:
 - A. methyl ethyl ketone
 - B. methyl propyl ketone
 - C. dimethyl ketone
 - D. diethyl ketone

Answer

- 40. The catalyst used for the polymerization of olefins is :
 - A. Ziegler-Natta catalyst
 - B. Wilkinson's catalyst
 - C. Pd-catalys
 - D. Zeise's salt catalyst

Answer

- 41. Bakelite is a:
 - A. urea-formaldehyde resin
 - B. phenol formaldehyde resin
 - C. polyethylene
 - D. artificial rubber

Answer

- 42. α -helix is found in :
 - A. DNA
 - B. RNA
 - C. Lipid
 - D. Protein

Answer

- 43. Aspirin is:
 - A. acetyl salicylic acid
 - B. 2-methoxy benzoic acid

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Answer



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44.

- A. fiuorobenzene
- B. benzene
- C. 1, 2-difluoro benzene
- D. 1, 3-difluoro benzene

Answer

- 45. Oxidation of aldehydes gives :
 - A. esters
 - B. acids
 - C. ethers
 - D. alcohols

Answer

- 46. The common acid used in the manufacture of rayon and plastic is :
 - A. methanoic acid
 - B. ethanoic acid
 - C. propanoic acid
 - D. butanoic acid

Answer

- 47. The compound used as an explosive is:
 - A. 2,4,6-tribromoaniline
 - B. 2,4,6-trinitrotoluene (TNT)
 - C. 1,3,5-trinitrobenzene
 - D. 2,4,6-trichlorotoluene

Answer

- 48. The indicator that is obtained by coupling the diazonium salt of sulphanilic acid with N, N-dimethyl aniline is
 - A. phenanthroline
 - B. methyl orange
 - C. methyl red
 - D. phenolphthalein

Answer